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A 4,5-DIOXOAPORPHINE FROM THE AERIAL PARTS OF STEPHANIA TETRANDRA

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ABSTRACT.—Chromatographic separation of the EtOH extract from the aerial parts of *Stephania tetrandra* resulted in the isolation of a novel 4,5-dioxoaporphine alkaloid, stephadione [1], together with six known alkaloids: corydione, oxonantenine, cassameridine, nantenine, cassythicine, and tetrandrine. The structure of stephadione was determined to be 6a,7-didehydro-1,2;9,10-dimethylenedioxo-4,5-dioxo-N-methylaporphine by a consideration of spectral evidence.

Many reports have dealt with the phytochemistry of the traditional Chinese medicine, Fen Fang Ji. This drug is the dried root of Stephania tetrandra S. Moore (Menispermaceae), which is widely distributed in Southern China. Twenty-one compounds, including nineteen alkaloids, have been determined (1-4). The aerial part of S. tetrandra appears to have escaped the attention of chemists. It was therefore thought desirable to investigate the constituents of the aerial parts of S. tetrandra. From the EtOH extract, one novel compound, stephadione [1], one well-known component, tetrandrine, and five other alkaloids previously unreported from S. tetrandra were isolated and purified.

The new optically inactive base, stephadione [1], was obtained as an amorphous red powder with mp $>300^\circ$.



¹Permanent Address: Department of Pharmacology, Jining Medical College, Jining 272113, People's Republic of China. Its hreims exhibited a molecular ion peak at m/z 349, corresponding to its molecular formula of $C_{19}H_{11}NO_6$ (found 349.0564, calcd 349.0582). The uv spectrum had five bands at 239, 271, 312.5, 326, and 466 nm and its ir spectrum showed the presence of a conjugated ketone at 1647 and a six-membered lactam at 1588 cm⁻¹. These data suggested that **1** was a 4,5-dioxoaporphine-type alkaloid.

In the ¹H-nmr spectrum of $\mathbf{1}$, there were signals for one N-Me group at δ 3.86 and two methylenedioxy groups at δ 6.21 and 6.64. The four protons in the aromatic region resonated as four singlets at 8 7.35, 7.61, 8.01, and 8.32, indicating that both rings A and D had a methylenedioxy group and the oxygenated pattern of ring D must be at 9,10. The bonded site of the methylenedioxy group in ring A was assumed to be at 1,2, because all the aporphine compounds previously reported from natural sources possess a 1,2oxygenation pattern (5-8). An nOe differentiation analysis further confirmed the bonded site of the oxygenation group in ring A. As no enhancement was observed through irradiation of H-11, which always appeared as a signal in the lowest field, the chemical structure of 1 was established as 6a,7-didehydro-1,2;9,10-dimethylenedioxo-4,5-dioxo-N-methylaporphine.

The six other compounds were shown to be the known alkaloids corydione (9), cassameridine (5), nantenine (5), tetran-

829

drine (4), and cassythicine (5), based on direct comparisons (mp, uv, ms, ¹H nmr, and co-tlc) with the respective authentic samples.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.— All mp's were uncorrected. Uv spectra were obtained on a Shimadzu UV-300 spectrophotometer with 95% EtOH as solvent. Ir spectra in KBr disk were recorded on Nicolet 5S XC and Perkin-Elmer 983 spectrometers. ¹H-nmr spectra were taken on JEOL FX-90Q and VARIAN XL-400 spectrometers in CDCl₃ with TMS as internal standard. Mass spectra were obtained on Nicolet FTMS-2000 and JMS DX300 spectrometers.

PLANT MATERIAL.—The aerial parts of S. tetrandra were collected in July 1988 in Jingdezhen, Jiangxi Province, China. A voucher specimen was deposited in the Phytochemistry Division, China Pharmaceutical University, Nanjing, China.

EXTRACTION AND ISOLATION. - Air-dried, ground, plant material was extracted exhaustively with boiling EtOH, and the extract was evaporated in vacuo to afford a residue which was extracted successively by refluxing with petroleum ether, CHCl₃, EtOAc, and MeOH. The CHCl₂ fraction was concentrated to give a gum which was treated with 2% HCl. The insoluble mass was subjected to a column of Si gel H with an elution gradient of C₆H₆ and Me₂CO to provide two chromatographically homogeneous components, corydione, and stephadione [1]. The aqueous acidic solution was fractionated between aqueous acid and CHCl₃. The upper layer was adjusted to pH 9 with NH4OH and extracted with CHCl₃. Both CHCl₃ extracts were chromatographed separately on columns of Si gel H. Stepwise elution of the columns with petroleum ether/CHCl₃, CHCl₃, and CHCl₃/MeOH afforded corydione, oxonantenine, cassameridine, and nantenine from the first column and cassameridine, tetrandrine, and cassythicine from

the second column.

Stephadione [1].—Amorphous red powder: mp >300°; CHCl₃ solution showed strong green fluorescence; uv λ max (EtOH) nm (log ϵ) 239 (4.54), 271 (3.99), 312.5 (4.13), 326 (4.47), 366 \pm 3 (4.13); ir ν max (KBr) cm⁻¹ 3077, 3029, 2916, 1647 (conjugated carbonyl), 1588 (lactam), 1499, 1467, 1363, 1305, 1247, 1191, 1079, 1045, 939; eims m/z [M]⁺ 349.0564 (78.3), 321 (100), 304 (31.1), 292 (16.2), 263 (19.8), 207 (8.0), 206 (62.1), 190 (14.2); ¹H nmr see text and structure.

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LITERATURE CITED

- M. Tomita, Yakugaku Zasshi, 87, 316 (1967).
- T.M. Hu and S.X. Zhao, Acta Pharm. Sin., 21, 29 (1986).
- T. Ogino, T. Sato, H. Sasaki, M. Chin, and H. Mitsuhashi, *Heterocycles*, 27, 1949 (1988).
- J.Z. Deng, S.X. Zhao, and F.C. Lou, J. Nat. Prod., 53, 993 (1990).
- H. Guinaudeau, M. Leboeuf, and A. Cave, J. Nat. Prod., 38, 275 (1975).
- H. Guinaudeau, M. Leboeuf, and A. Cave, J. Nat. Prod., 42, 325 (1979).
- H. Guinaudeau, M. Leboeuf, and A. Cave, J. Nat. Prod., 46, 761 (1983).
- H. Guinaudeau, M. Leboeuf, and A. Cave, J. Nat. Prod., 51, 389 (1988).
- H.G. Kiryakov, E. Iskrenova, B. Kuzmanov, and L. Evstatieva, *Planta Med.*, 43, 51 (1981).

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